

EngrD 2190 – Lecture 19

Concept: Process Analysis - Graphical Modeling

Context: Thermodynamic Maps for Binary Mixtures -
The Lever Rule and Tie Lines

Defining Questions:

What is the descriptive term for the border between the ‘all vapor’ and ‘liquid+vapor’ regions on a map for binary mixtures?

What is the descriptive term for the border between the ‘all liquid’ and ‘liquid+vapor’ regions on a map for binary mixtures?

Bring a Straightedge or Ruler to Lecture 20.

EngrD 2190 - Second Team Assignments - Fall 2025

Team 1	NetID
Gabby Alvarado	gaa68
Valerie Ruiz	vr336

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Chuck Fang	cf544
Cullen McCarthy	cpm227

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Em Vincent	ekv24
Sophia Zheng	sz683

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Andy Glesmann	arg292
Bradley Reiff	bjr235

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Kanat Schoeberlein	kts59

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Bill Nguyen	nn347
Wilson Zhang	wtz5

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Andreas Custelcean	aec327
Jude Thomas	jct268

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Sofia Romero	scr226
Shivani Parmar	ssp238

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Connor Lucente	cbl64
Wilman Vlach	wlv23

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Steffanie Jones	scj56
Anna Voronova	asv44

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Hank Xue	zx344
Xinchi Zhang	xz985

Team 13	NetID
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Syon Khatter	sk3297
Vanshikha Sabbi	ls2242

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Daniel Herzenstein	dh762
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Dolly Hritz	drh258
Audrey Fu	af643

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Nadia Diaz	nld46
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Evelyn Sangree	ehs95

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Patrick Donahue	pnd25

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Malvika Rao	msr298
Besi Santana	bs862

Team 28	NetID
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Jasmine Hwee	jh2875
Maddie Li	ml2825

Team 29	NetID
Lila Chami	lc2259
Lila Kaman	lk567
Lizeth Montemayor	lm945

Team 30	NetID
Tanbeer Islam	ti232
Josh Lennon	jsl379
Ayush Tripathi	at969

Homework

- Practice Exercises (not to be handed in):
Process Economics **3.96** and **3.97**. Solutions are posted.

- Homework 6 due Friday 10/24.
4.4 and **4.8** graphical energy balances.
4.15 graphical analysis of flash drums

Download blank graphs and phase data from EngrD 2190 homepage:

Textbook → Textbook Graphs and Figures →
Graphs for Chapter 4 Exercises.

Homework is your chief means of assessing your command of the material.

- New Homework Team Assignments
For New Teams: **Last** person listed is Team Coordinator for Homework 6.
- Process Economics Analysis with a Spreadsheet
Optimize a chemical process with respect to ROI.
Your *Process Plan* – equipment purchases and operating parameters –
is due Monday 10/20 at 5 p.m.
You must also submit your spreadsheet.

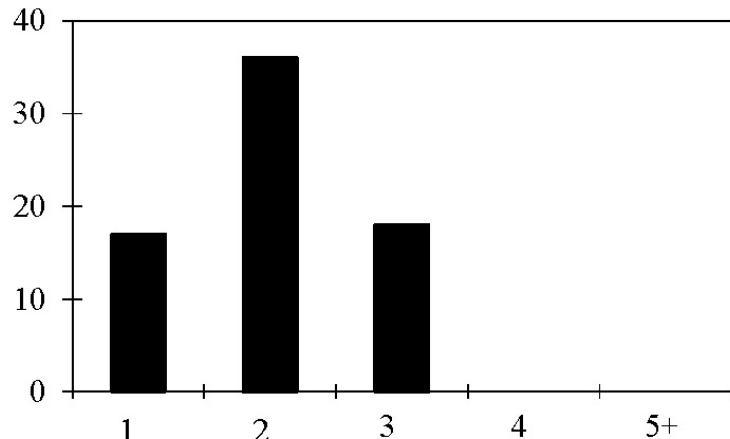
Mid-Semester Survey - 2025

EngrD 2190 - Chemical Process Design & Analysis - Fall 2025

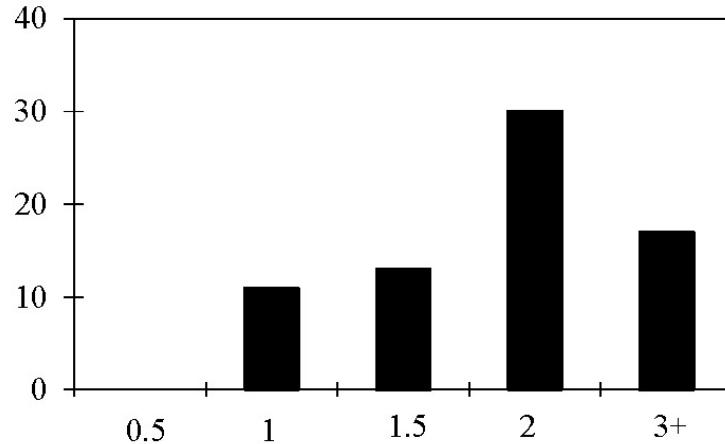
86 enrolled, 72 responses

Team Homework

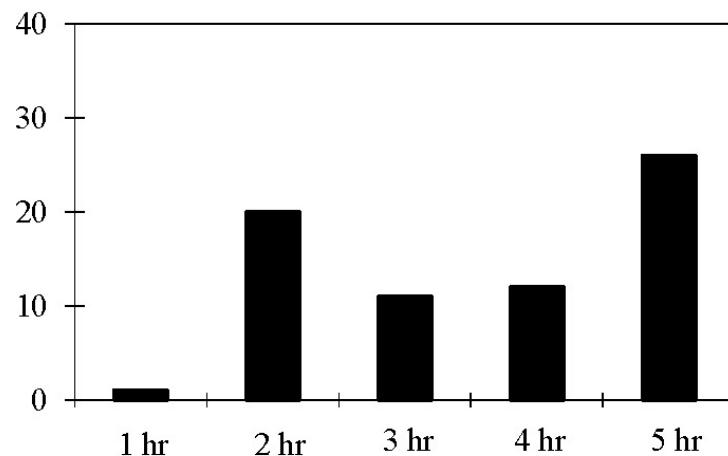
How many times does your team meet to complete a homework assignment?



What is the duration of a typical team meeting (in hours)?



Total time in meetings weekly?

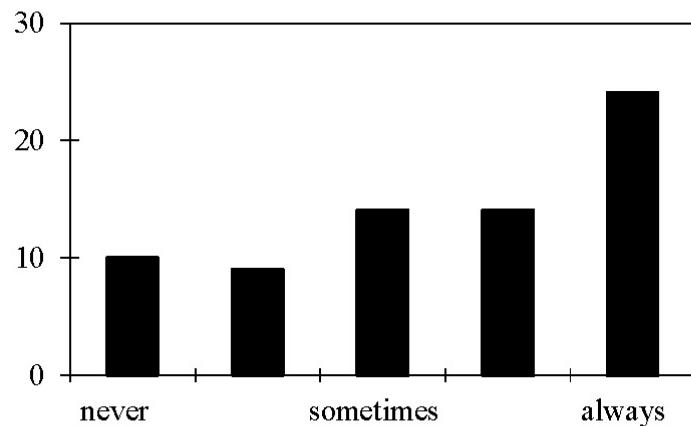


Mid-Semester Survey - 2025

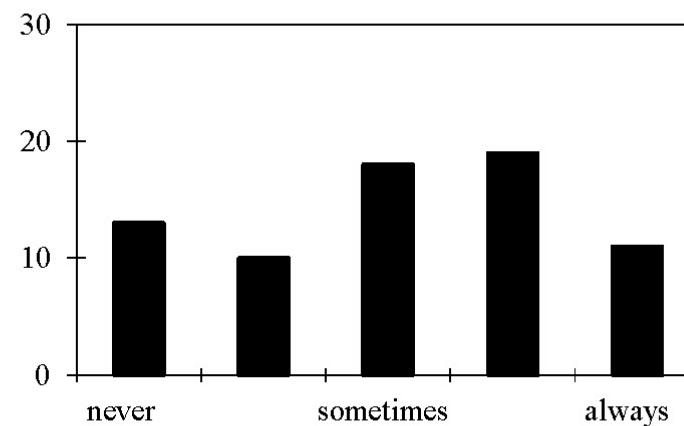
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Team Homework, continued

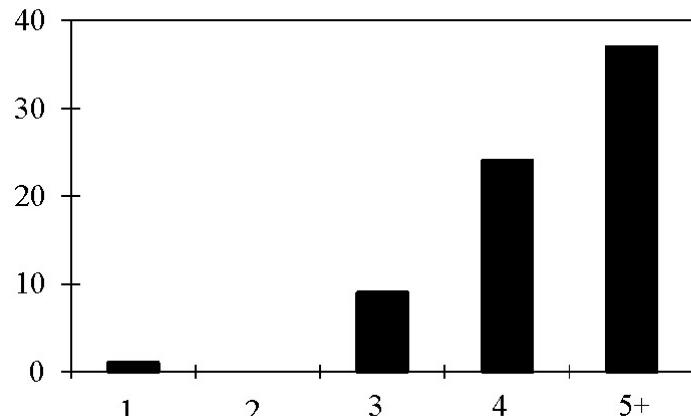
Do you work the homework exercises
before your team's first meeting?



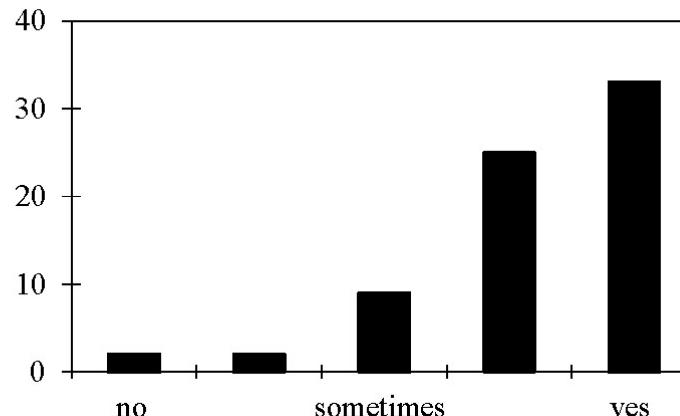
Do your teammates work the exercises
before your team's first meeting?



What is your total time spent
on a typical homework assignment (hours)?



Is your homework team effective
for learning the course material?

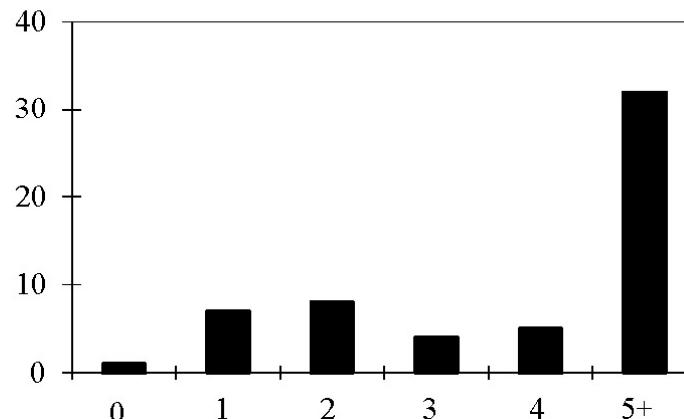


Mid-Semester Survey - 2025

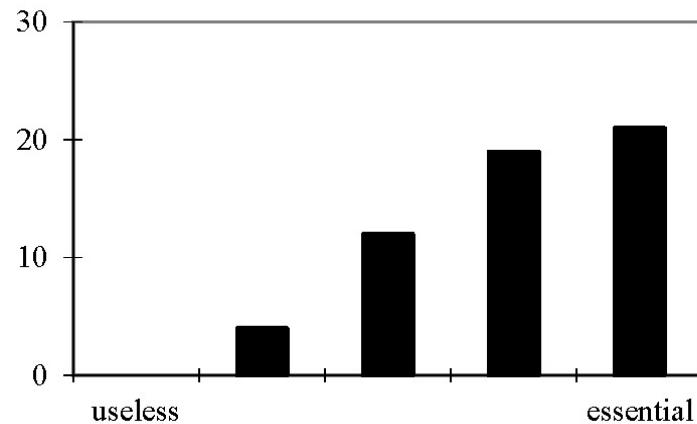
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86 enrolled, 72 responses

TA Office Hours

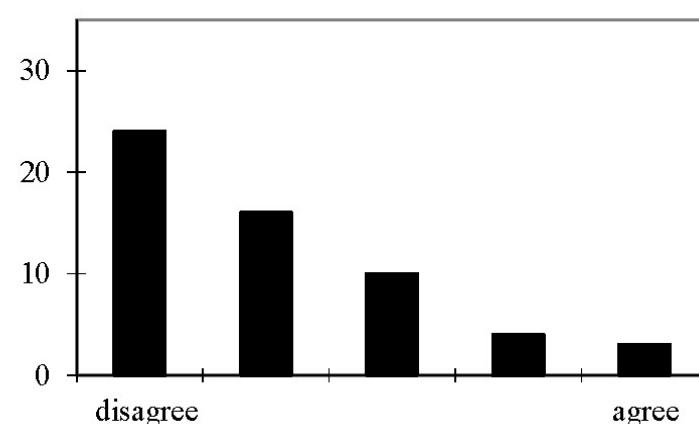
How many times have you visited TA office hours?



If you have consulted the TAs during office hours, please rate the usefulness.

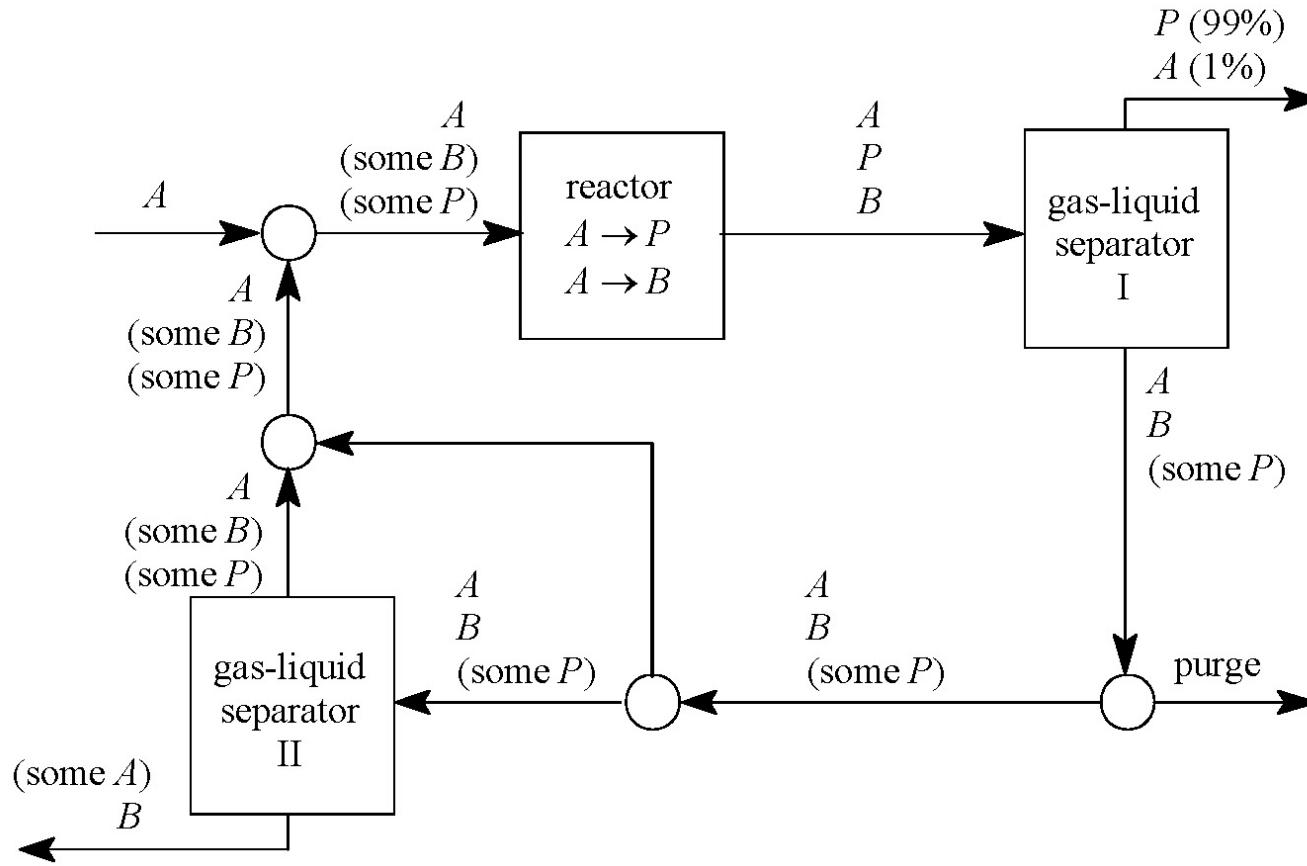


Two TAs are sufficient to cover the demand in TA office hours.



Process Economic Analysis with Spreadsheets - 2025

Optimize a chemical process with respect to ROI.



Annual Plans due Monday 10/20 ~~at noon~~ **at 5 p.m**

Process Design:

Equipment Capacities?

Process Operation:

Feed flow rate? Purge fraction? Recycle to Reactor?

Recycle through 2nd Separator?

Process Economic Analysis with Spreadsheets - 2025

Process Design & Operation Plan - 2025

Name _____
GIVEN NAME _____

Equipment Purchaser _____

Reactor: _____

Separator: _____

Separator II: Type: _____

Operating Parameters: _____

Reactor: _____

Frac. Recycle: _____

Fraction of Separator I liquid bottoms to recycle*: _____

All eight configurations
can be optimized to a $ROI > 0$.

The maximum ROI is 36%/year.

The minimum ROI is 18%/year.

You should be able to attain
(1 or 2) a ROI of at least 18%/year.

You should be skeptical if you attain
a ROI greater than 36%/year.

Types
dictated
by
your
NetID

Download
this form
from the
EngrD 2190
website

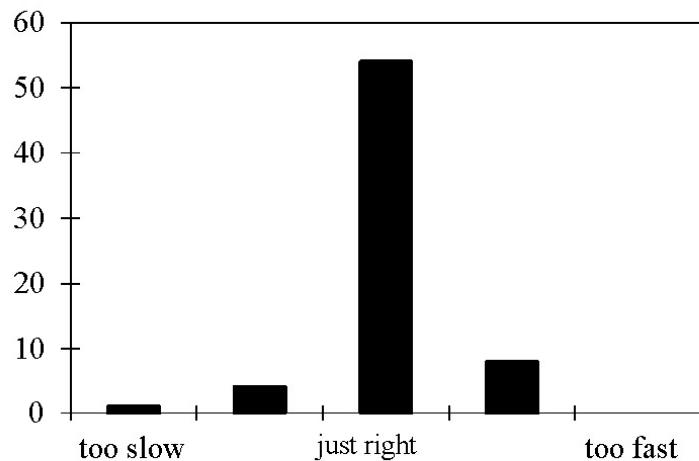
Due Monday October 20 ~~at noon~~ at 5 p.m

Mid-Semester Survey - 2025

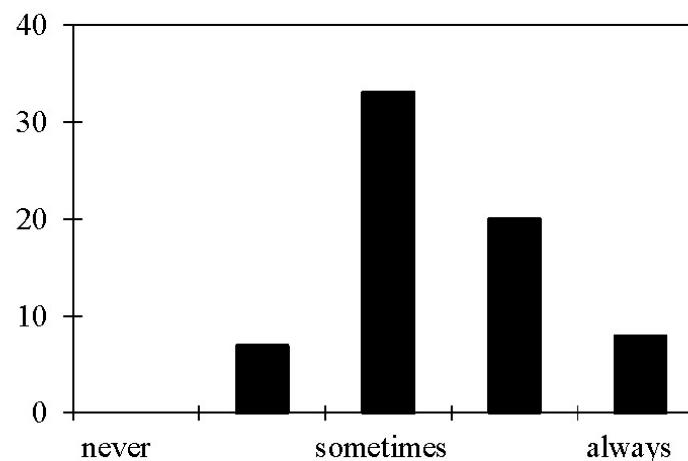
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Lectures

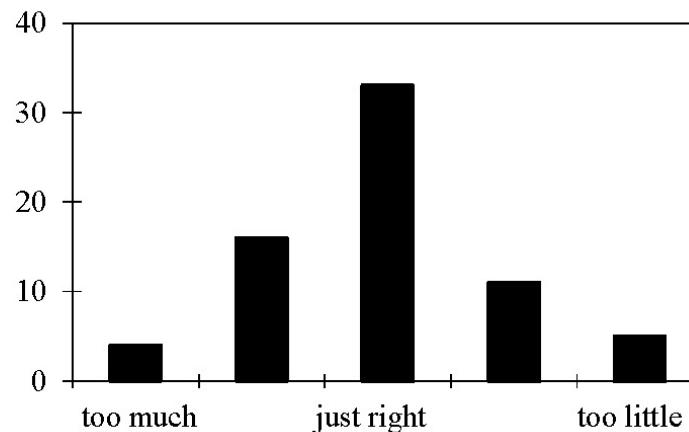
Pace of the lectures?



Read the textbook before the lecture?



Overlap between the lectures and the textbook?

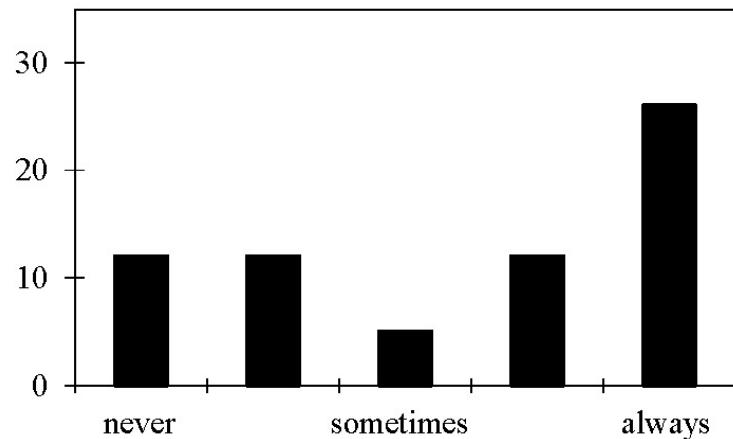


Mid-Semester Survey - 2025

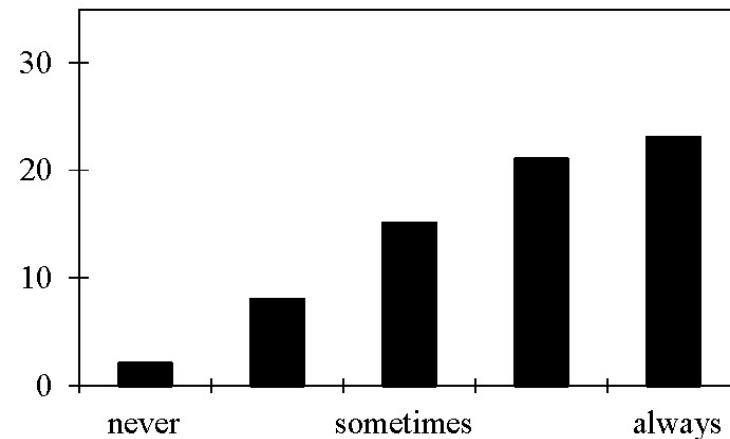
EngrD 2190 - Chemical Process Design & Analysis - Fall 2025
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Lectures

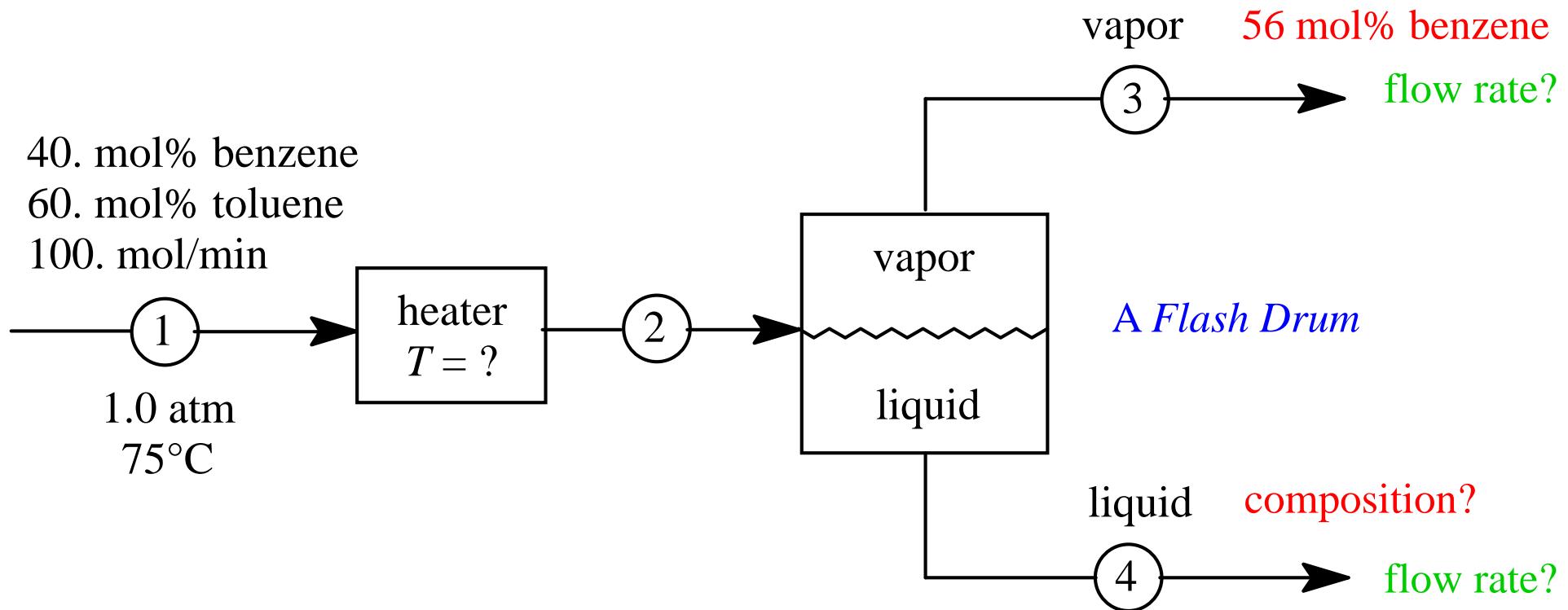
Review the posted powerpoints and pdf's of the lecture ***before or during*** the lecture?



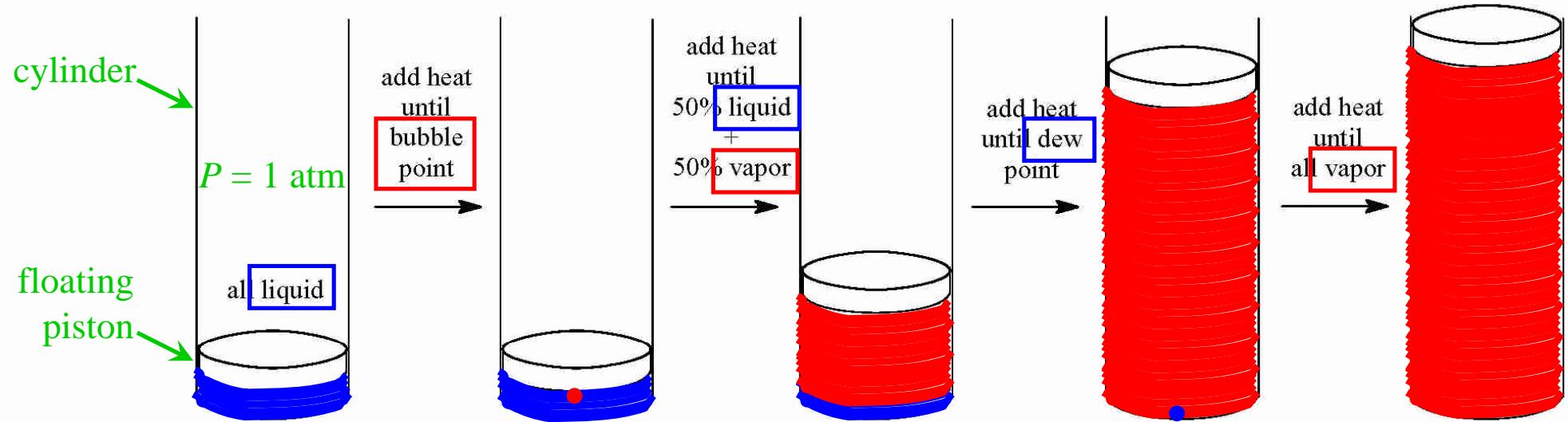
Review the posted powerpoints and pdf's of the lecture ***after*** the lecture?



One L+V Equilibrium Stage

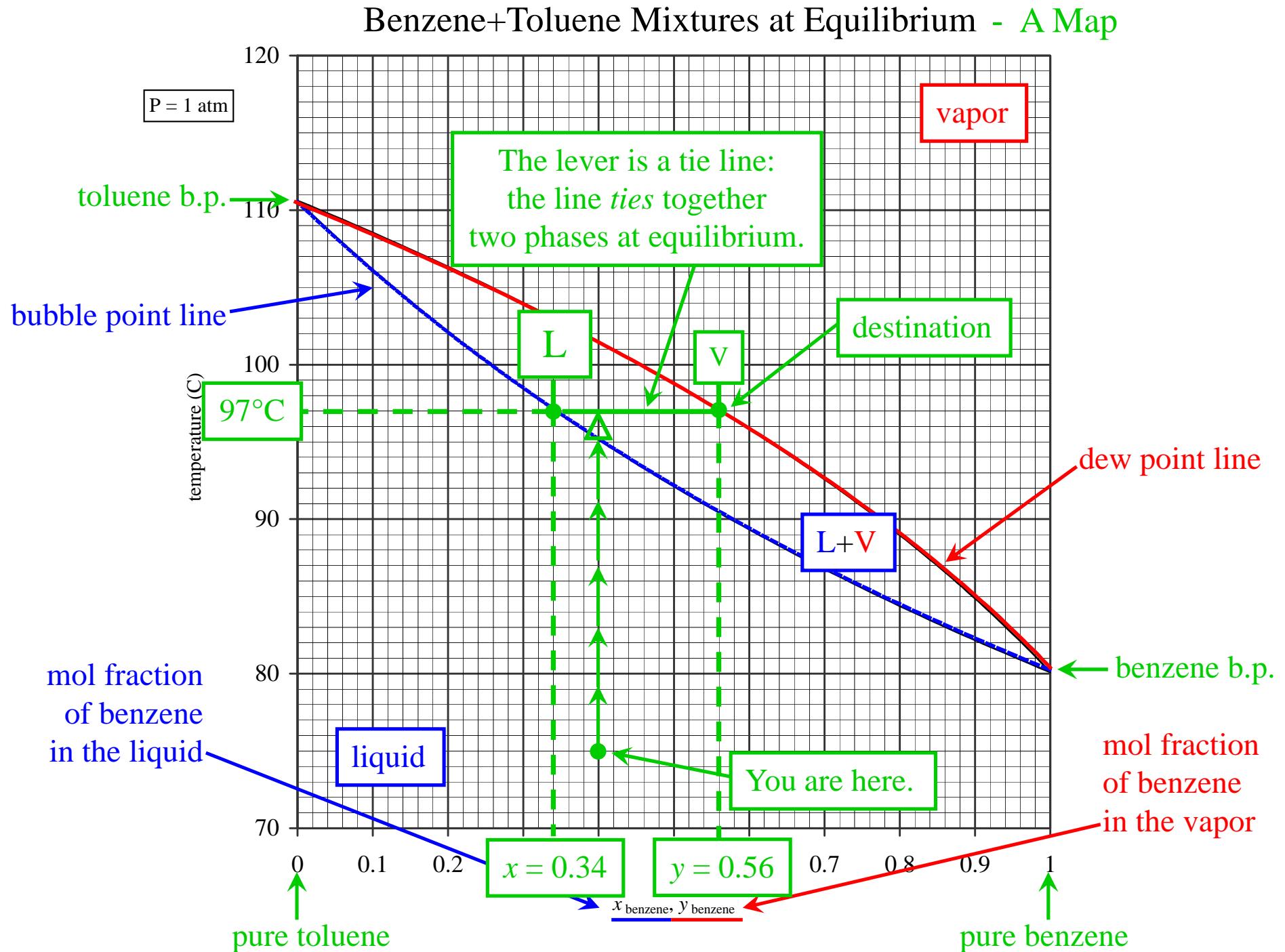


Binary Mixtures: Phase Changes by Increasing T at Constant P



	$T < 80^\circ\text{C}$	80°C	80°C	80°C	$T > 80^\circ\text{C}$
benzene	$T < 80^\circ\text{C}$	80°C	80°C	80°C	$T > 80^\circ\text{C}$
toluene	$T < 111^\circ\text{C}$	111°C	111°C	111°C	$T > 111^\circ\text{C}$
50% benzene 50% toluene	$T < 92^\circ\text{C}$	92°C L: 50% B V: 70% B	95.5°C L: 38% B V: 62% B	98.5°C L: 30% B V: 50% B	$T > 98.5^\circ\text{C}$
75% benzene 25% toluene	$T < 85.5^\circ\text{C}$	85.5°C L: 75% B V: 88% B	87.5°C L: 66% B V: 84% B	91°C L: 54% B V: 75% B	$T > 91^\circ\text{C}$
25% benzene 75% toluene	$T < 100^\circ\text{C}$	100°C L: 25% B V: 45% B	103.5°C L: 17% B V: 33% B	105°C L: 11% B V: 25% B	$T > 105^\circ\text{C}$

Example 1.



Example 1 - Apply the Lever Rule to the Tie Line



$$\text{total lever: } 0.56 - 0.34 = 0.22$$

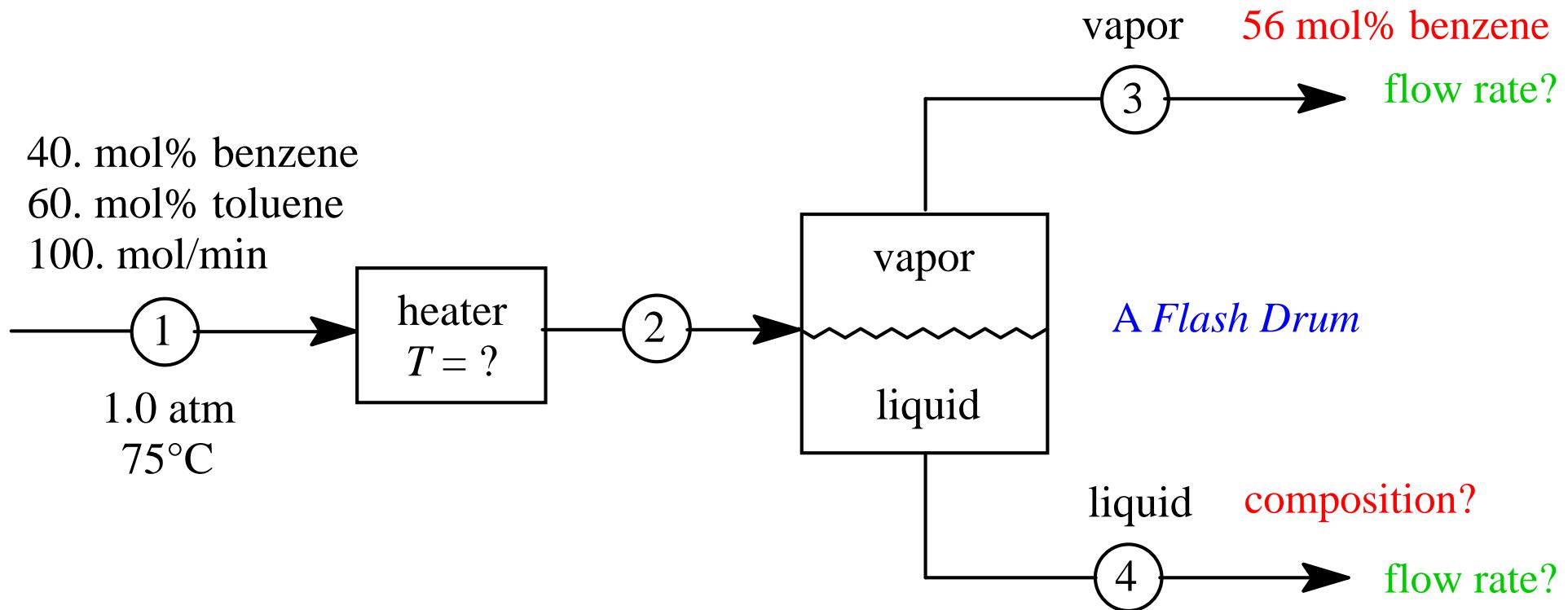
$$\text{lever arm to liquid: } 0.40 - 0.34 = 0.06$$

$$\cancel{\text{liquid flow rate}} = 100 \times 0.06 / 0.22 = 27 \text{ mol/min.} \quad \text{Okay?}$$

vapor

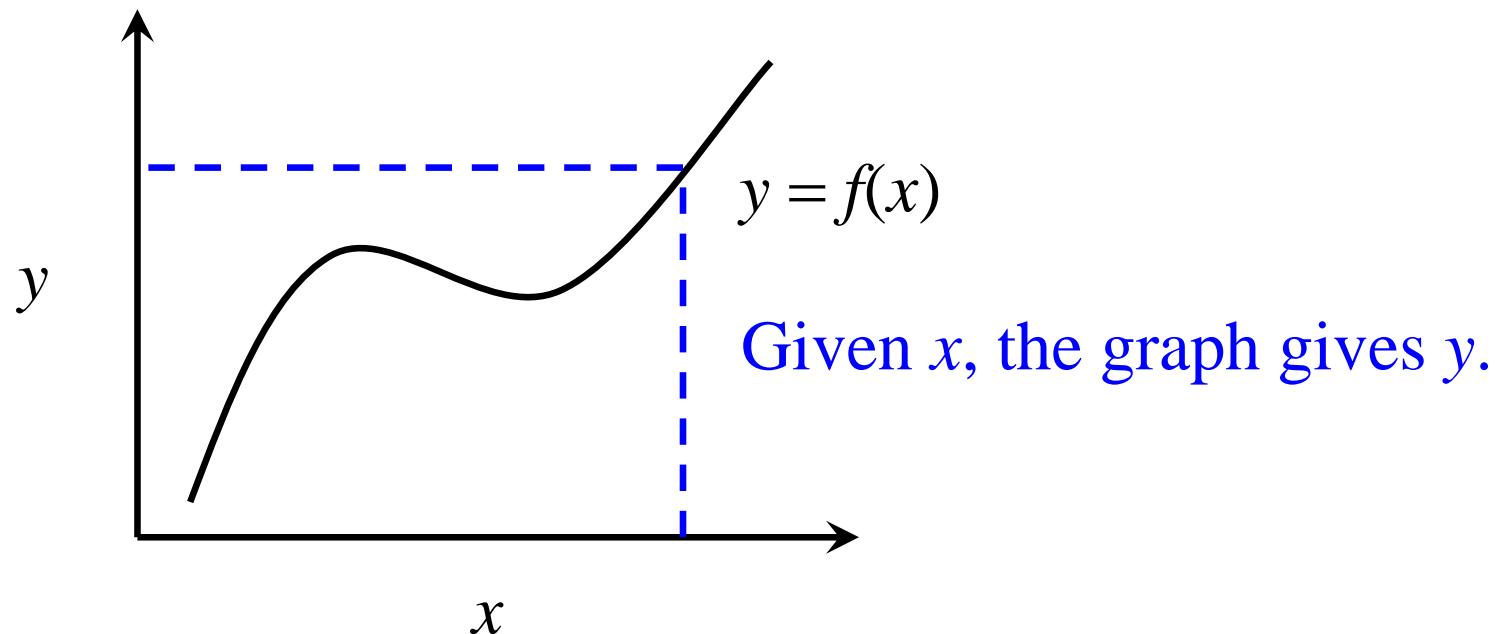
The following five slides
were developed simultaneously
on the second screen.

One L+V Equilibrium Stage



How to plot these Data? A Graph?

A Graph:

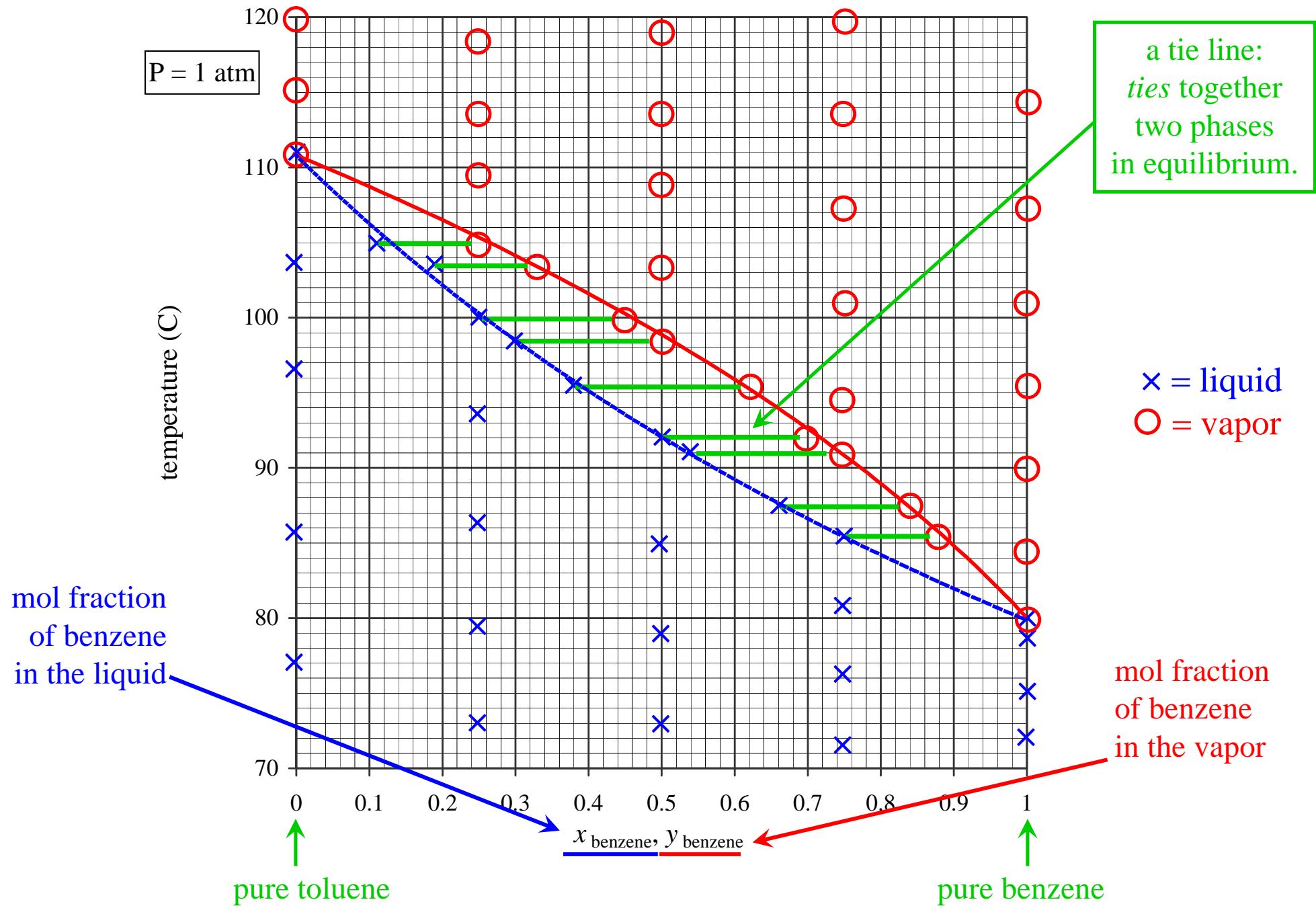


Our data: given composition *and* temperature,
what is the phase (liquid, vapor, or liquid+vapor)?

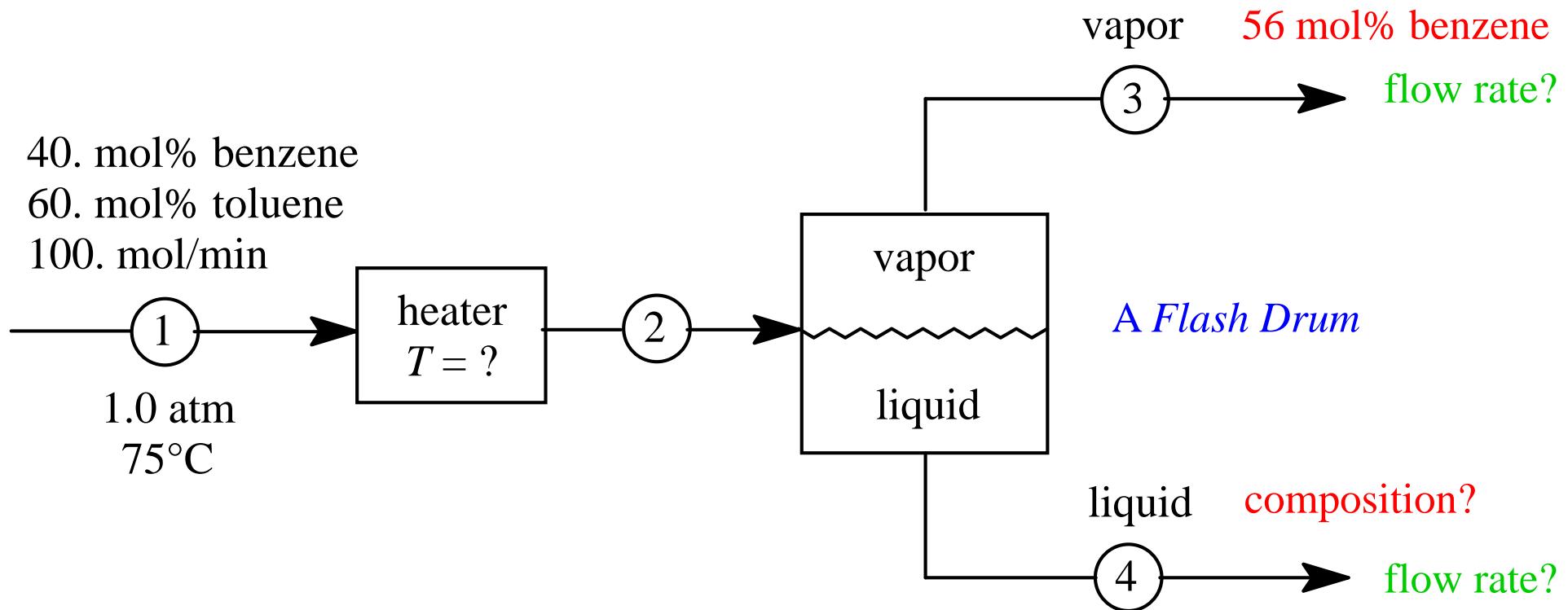
Two independent variables - composition *and* temperature.

We need a *map*.

Benzene+Toluene Mixtures at Equilibrium A Map

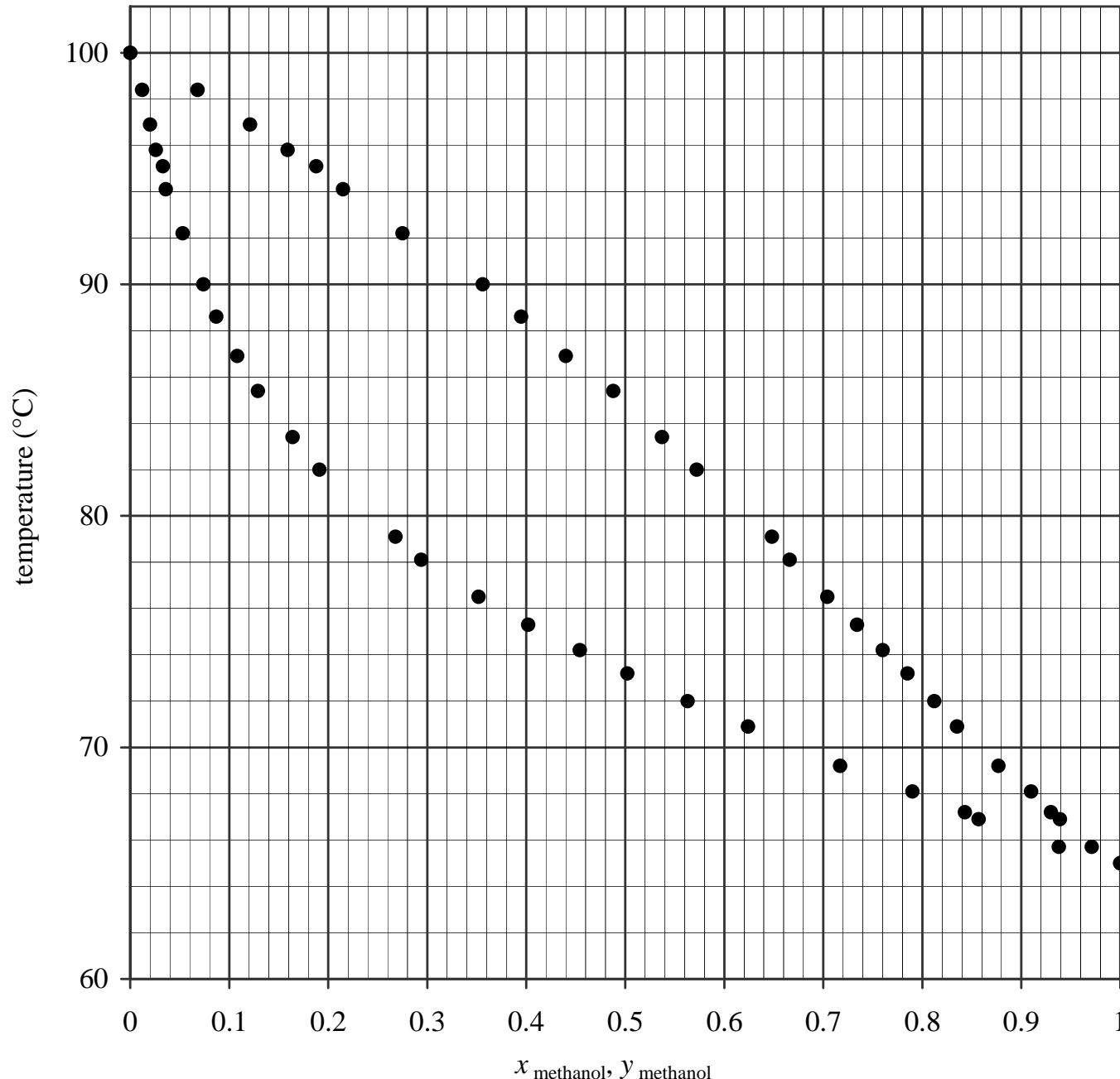


One L+V Equilibrium Stage



The following two slides
were not used Fall 2025.

Vapor-Liquid Phase Map for Methanol + Water at 1 atm - Exercise 4.19



Vapor-Liquid Phase Map for Methanol + Water at 1 atm - Exercise 4.19

Do not
'connect the dots.'

Sketch 'best fit'
lines or use
French curves.

